

Chapter 9 Review Stoichiometry

Section 2 Answers

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Section 3 Answer Key Chapter 9 Review Stoichiometry Section 3
Answer Key

What is Stoichiometry? ... (NO₃)₂(s) + 2Ag(s)Solve. Answer. 5.93 x
10²⁴ atoms of Ag are produced from 275.0 g of Fe. Solving problems
with mixed units. Stoichiometry knowledge is required to solve
problems caused by everyday chemistry in unique situations. ... Chapter
9: Section 2...

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work in the space provided. 1. The following equation represents a
laboratory preparation for oxygen gas: $2\text{KClO}_3(\text{s}) \rightarrow 2\text{KCl}(\text{s}) + 3\text{O}_2(\text{g})$
How many grams of O₂ form if 3.0 mol of KClO₃ are totally
consumed? 2. Given the following equation: $\text{H}_2(\text{g}) + \text{F}_2(\text{g})$

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— LCi(,; — h. The oxygen gas produced in part a has density of 1.43

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CHEMISTRY a. —. 81 g 6. A car air bag requires 70. L of nitrogen gas
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PROBLEMS Write the answer on the line to the left. Show all your
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CHAPTER 9 REVIEW Stoichiometry SECTION 3 PROBLEMS Write
the answer on the line to the left Show all your work in the space
provided 1 88% The actual yield of a reaction is 22 g and the theoretical
yield is 25 g Calculate the percentage yield 2 60 mol of N_2 are mixed
with 120 mol of H_2 according to the following equation: $\text{N}_2(\text{g}) + 3\text{H}_2(\text{g})$

SECTION 2 continued Date Class _____ 60.2 9 42.1 1 a. \ tt mash 01

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— LCi(,; — h. The oxygen gas produced in part ahas density ot 1.43
g/L calculate the volume of this gas.. 76 STOICHIOMETRY MODERN
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1. 2. Mole Ratios a. Mole ratios are conversion factors that relate the number of moles of one chemical to 0.0209 mol AgNO₃ Aspirin production a. 13.5 kg C₉H₈O₄ b. 7.66 kg C₄H₆O₃ c. 4.29 L HC₂H₃O₂
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Chapter 9 Review Stoichiometry Section 2 Answers 9-1 Introduction

to Stoichiometry Composition Stoichiometry - deals with mass relationships of elements in compounds Reaction Stoichiometry - Involves mass relationships between reactants and products in a chemical reaction I. Page 4/15

Chapter 9 Intro to Stoichiometry Section 9.1 . Chapter 9 9.1 Objectives
• Define stoichiometry. • Describe the importance of the mole ratio in stoichiometric calculations. • Write a mole ratio relating two substances in a chemical equation. ... Chapter 9 9.2 Objectives

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